APPLICATION OF CERAMIC PREFORMS TO THE MANUFACTURING OF CERAMIC – METAL COMPOSITES

This paper presents new solution to manufacturing Cu-Al$_2$O$_3$ composites with a dispersed reinforcement phase. It consists in substituting commonly used commercial alumina powder with crushed ceramic foam. Very low strength of thin-walled Al$_2$O$_3$ porous foams results in both their easy destruction and isolation of fragments $<200$ µm. The addition of the ceramic phase is 3-10 vol.%. The sintering process is performed for 30 min in a hot-press at 1050°C and 30 MPa. The aim of the presented paper is to study the effect of the volume content of the ceramic phase on the microstructure and selected physical and thermal properties of copper – alumina composites.

Keywords: ceramic preform, composite, copper, sintering

1. Introduction

Alumina/copper composites are used when high thermal conductivity, high absorption and dissipation of heat, high resistance to thermal fatigue and good frictional wear resistance are required. Their properties depend on a number of factors, including the content, shape and distribution of the ceramic phase, the method of its preparation as well as the conditions under which they are fabricated. All these variables have strong influence on the above mentioned characteristics and, in consequence, on the future applications of the final material.

In this paper the preparation method of copper–alumina composites will be presented. Dispersion-strengthened Cu-Al$_2$O$_3$ composites are extensively used as materials for applications which require high-strength and good electrical conductivity[1]. Ceramic/metal composites can be produced by a number of methods, i.e. direct oxidation of most appropriate metal [2], metal infiltration (pressure or pressureless) of porous alumina foams [3-5], mechanical alloying [9-11], explosive compaction [12], in situ processing [13,14], spark plasma sintering [15], etc.

The most basic technique for producing copper-alumina composites includes mixing two powders in suitable proportions and sintering them. However, it has one considerable drawback. Aluminium oxide added in powder form is agglomerated and does not sinter at the temperature at which the composite is prepared, which is determined by the melting point of copper. This article provides an efficient solution to this problem, which consists in using a fragment of the sintered ceramics derived from foams as the dispersed phase instead of alumina powder.

Studying the effect of using different amounts of ceramic phase additives on the microstructure and selected physical and thermal properties of copper – alumina composites is among the aims of the present investigation.

2. Experimental

The preparation procedure of ceramic foams was described in more detail in the previous work [16]. In short, ceramic foams were fabricated following the replication method. The ceramic powder, solvent, dispersant and binder were mixed in a Pulverisette planetary mill. The ceramic slurry thus prepared was deposited on polymeric sponge. After sintering, we obtained a ceramic material with the structure that was an ideal copy of the structure of the polymeric substrate (Fig. 1).

In this work, the Al$_2$O$_3$ content in the slurry was 75.0 wt.%,
whereas the addition of the acrylic-styrene dispersion was 5 wt.% of the mass of the ceramic powder.

![Fig. 1. The microstructure of sintered ceramic foam (SEM)](image1)

Very low strength of the structure allowed its easy destruction and isolation of a fraction with the size of < 200 µm. Next, the fabrication of composites began with the use of ceramics derived from the ceramic preform. The morphology and size distribution of the ceramic phase is shown in Figure 2. Ceramics obtained after a sieve analysis have different particle size fractions (2-400 µm), with the mean size of 46 µm.

![Fig. 2. Sintered and shattered Al₂O₃ preform; (A) SEM image, (B) distributions of the size](image2)

The first stage consisted in preparing a mixture of metal (copper powder <100 µm, 99.9% purity) and ceramics. The addition of the ceramic phase was 3.0 vol.%, 5.0 vol.% and 10.0 vol.%. In order to obtain a mixture of the highest possible homogeneity, mixing was performed in a planetary mill (Pulverisette 6, Fritsch) with tungsten carbide balls (Ø 5 mm). High-energy mechanical milling experiments were performed at the room temperature under air atmosphere, at the rotation speed of 200 rpm and at the time of mixing 6h. The weight ratio of balls to powder (BPR) was 3:1. The sintering process of composites ran for 30 min in a hot-press (ASTRO, Thermal Technology, USA) at 1050°C and 30 MPa in an argon atmosphere.

3. Methods

The morphology and microstructure of the composite materials were examined using a scanning electron microscope (SEM AURIGA CrossBeam Workstation Zeiss). The characterization of the size distribution of the ceramic fraction was performed employing a CLEMEX television image analysis system. The relative density and open porosity were measured according to the Archimedes' principle. The hardness (HV5) was tested by Durascal 10/Emcotest with a Vickers diamond indenter using load 9.81 N with the loading time of 10 s. Thermal diffusivity was measured within the temperature range 50-600°C using the Laser Flash Analyser LFA457/Netzsch. Tribological tests were performed by using a DUCOM tribotester, in a “ball – flat surface” friction mode. The samples were pressed against a stainless steel ball with 6.5 mm in diameter with force ca. 10 N. After the test, the surface of the groove was analysed using scanning electron microscopy. The friction force thus generated was measured with a piezoelectric displacement sensor 24 times per second. It was induced and recorded during 30.0 min long friction processes.

4. Results and discussion

The microstructure of composite materials is shown in Fig. 3. As can be seen, the resulting composites are characterized by a homogeneous distribution of the reinforcing phase (Fig. 3A-C); however, the difference in size is significant. As a result of the conducted analysis, a strong binding was reported between the copper matrix of the composite and its ceramic phase (Fig. 3F).

![Selected properties (physical and thermal) of Cu-Al₂O₃ composites produced by hot pressing (HP) are presented in Table I. It is worth noting here that by mixing the two phases of composites, very high densities, close to the theoretical ones, were obtained. The composites contained both many fragments of the size of about 100 µm (Fig. 3D), as well as fragments smaller than 10 µm (Fig. 3E). The reinforcement ceramic phase came in a wide range of sizes. Despite this, a trend of increased hardness of materials was clearly observed when increasing the alumina content. The hardness of](image3)
the composite containing 3 vol.% Al$_2$O$_3$ was approx. 48 MPa, whereas that of the composite containing 10 vol.% of ceramic equaled ca. 87 MPa.

Fig. 3. Sample image (SEM) of the microstructure of copper composites with Al$_2$O$_3$ from foams; A) 3 vol.% Al$_2$O$_3$, B) 5 vol.% Al$_2$O$_3$, C) 10 vol.% Al$_2$O$_3$, D) larger ceramic fragments in the composites, E) the smallest ceramic fragments in the composite, F) a phase boundary

Fig. 4. The thermal conductivity of Cu-Al$_2$O$_3$ composite materials vs. the temperature

Thermal properties of the composites are shown in Fig. 4. The produced composites are characterized by high thermal conductivity (330 W·m$^{-1}$·K$^{-1}$ for Cu – 3 vol.% Al$_2$O$_3$), which decreases with increasing both the content of the ceramic phase (260 W·m$^{-1}$·K$^{-1}$ for Cu – 10 vol.% Al$_2$O$_3$), as well as the temperature. This behaviour is typical for this type of materials.

The abrasion resistance of the composites was also tested. Changes in the coefficient of friction diagrams presented in Figure 5 indicate that after several cycles of friction (equivalent to a test lasting approximately 20-50 s) the friction coefficient stabilizes at 1.3-1.4. This value is much higher than in the case of composites prepared analogously with the use of commercial alumina powder as the ceramic phase, for which the friction coefficient of 0.3-0.4 was obtained. At the same time, regardless of the content of the ceramic phase in the composite, the profiles of traces are very similar in size and shape, as shown in the images presented in Fig. 6.

Fig. 5. The friction coefficient of copper-alumina composites

In general, there are no significant differences between the tribological properties of materials with the Al$_2$O$_3$ content ranging from 3 to 10 vol.%. Probably, the greatest discrepancy can be observed when the ceramic phase content in composites is even higher. We will focus on such composites in the next phase of our investigation.

5. Conclusions

Alumina derived from crushing of foams used as the ceramic phase allowed obtaining composite materials with advantageous properties. As shown, foams used in this study were homogeneous and their microstructure was free of voids. That is how porosity was eliminated in the composite in the area of the reinforcing phase, which is unattainable during the manufacture of composites using alumina powder which undergoes agglomeration. Additionally, using low pressure (30 MPa) when producing the composite resulted in obtaining the relative density of approximately 98-99%. This translated into other measured values. Materials with high thermal conductivity of 330 W·m$^{-1}$·K$^{-1}$ for Cu – 3 vol.% Al$_2$O$_3$ were obtained. The higher the content of the ceramic phase, the greater the hardness of the composites (for Cu – 10 vol.% Al$_2$O$_3$ is close to 90 MPa). The resulting composites have high resistance to abrasion. Regardless of the type of the composites, the friction coefficient is in the range 1.3-1.4.
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