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1. Introduction

Semi-Solid Metal processing (SSM) of steel is a method of producing near-net shape products [1]. This technology utilizes the thixotropic flow of metal suspension [2]. Metal alloys are suitable for thixoforming if they have globular microstructure in a wide solidus-liquidus range [3]. The thixo-route is a two-step process, which involves the preparation of a feedstock material of thixotropic characteristics and reheating the feedstock material to semisolid temperature to produce the SSM slurry, subsequently used for component shaping [4]. The goal of the applied thixoforming technology is to improve the properties of elements manufactured in one-step operation, ensuring their high structure integrity, superior to casting and similar to that of wrought parts. Nowadays, rheoforming and thixoforming of Al-based and Mg-based alloys are commonly applied in the industry [3-6]. In the case of high melting alloys, such as steels, this technology is in the implementation stage [4]. Several kinds of steels e.g. 100C6, X210CrW12, M2, C38, C45, 304 stainless steels have been investigated for the application for SSM processing [4-9, 11-13]. A globular microstructure is mainly obtained by Strain Induced and Melting Activation (SIMA), Recrystallization and Partial Melting (RAP) and the modification of chemical composition by grain refiners methods [4, 10-12].

2. Thixoforming procedure

The thixo-casts were produced using a specially build prototype device. A piston velocity of 1 m/s was applied. The locking force of the machine was 800 kN [14]. A billets (diameter – 30 mm, height – 30 mm) of 100Cr6 [12, 14], X210CrW12 [11] steels and 7075 aluminum alloys with additions of Sc and Zr [10] were placed in the coil of an inductive heating furnace at 1425 °C, 1250 °C and 632 °C respectively. The temperature of the feedstock was measured with S type thermocouple. The billet was then moved to a shot sleeve of a high-pressure diecasting machine and forced by a piston into pressing force 35 kN the die, made of M2 steel, pre-heated to about 180°C and sprayed with BN [14].

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3. Characterization of 100Cr6 semi-solid range and microstructure after SSM

The 100Cr6 bearing steel used was produced by HSW Quality Steel Works. The steel was hot forged at approximately 1050°C. Next it was normalized at 880°C. The chemical composition was 0,9% C, 1,4% Cr, 0,4% Mn, 0,3% Si, 0,2% Cu (all in weight %). The curve in Figure 1 shows the dependence of the amount of liquid phase as a function of temperature for the 100Cr6 steel calculated from the DSC measurements. It shows melting start at 1307°C, and the end of melting process at 1497°C. Semi-solid processing was carried out at 25% of liquid phase which corresponded to the temperature of 1425°C [12].



Fig 1 DSC heating curve for 100Cr6 steel after forging [12]

Thixoforming of 100Cr6 steel was carried out according to the procedure described in paragraph 2. Figure 2 shows an optical microstructure of a thixo-cast sample. The microstructure consisted of a primary globular grains (average size 343 μ m) and secondary fine grains formed from a liquid phase during cooling (average size 20 μ m) [12].



Fig 2 Optical microstructure of 100Cr6 steel after thixoforming [15]

4. Characterization of feedstock and solidus-liquidus range of X210CrW12 tool steel The X210CrW12 steel with a high content of chromium and carbon belongs to the ledeburite class of steels. Figure 3 shows an light microstructure in perpendicular direction to rolling.

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It shows a primary carbide precipitation in the ferrite/pearlite matrix formed in bands parallel to the working direction. The average hardness of the billet is 220 HV_5 . Such a steel was used as a feedstock for the thixoforming process [11].



Fig 3 Microstructure of X210CrW12 steel after rolling – feedstock for thixoforming [15]

In Figure 4 the DSC curve (1) shows the endothermic effects, which occurred during heating at the rate of 20° C/ min. The curve (2) in Figure 4 shows the dependence of the calculated amount of liquid phase as a function of temperature. The melting point is determined as the onset of the endothermic peak.



Fig 4 DSC heating flow and liquid fraction curves for X210CrW12 steel after rolling [11]

The start of melting is set at such a temperature when the heat curve falls away from the tangent line. The end of melting is set by the onset point on the other side of the peak. A strong endothermic effect begins at 1209°C and ends at 1379°C. The enthalpy of the melting process is 612 J/g. The analysis of the melting curve shows that in the range 1209°C - 1256°C (0-42% liquid phase) the enthalpy of melting is much larger than during the rest of the process. This effect is connected with melting of the eutectic. Over 1256°C mainly solid solution melts. This effect is also observable in curve (2), which shows different kinetics of changes of the amount of liquid phase as a function of temperature.

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Semi-solid processing was carried out at 1250°C which corresponds to 38% of liquid phase in accordance with the DSC analysis (point A marked in Fig 4).

4.1. Microstructure of X210CrW12 after SSM

The optical microstructure of a X210CrW12 thixo-cast cross-section is shown in Figure 5. Visible austenite globular grains are surrounded by the eutectic mixture. The insert in Fig.5 shows a histogram of grain size. The distribution of grain size is within 15 μ m to 90 μ m. The largest fraction belongs to the grains in the range of 39 μ m to 48 μ m, indicating the average size of globules – 44 μ m. The average hardness of the thixo-casts was 401 HV₅



Fig 5 Microstructure of the thixo-casts made of steel X210CrW12 and quantitative analysis of grain size distribution [11]

5. Semi-solid range of Al7075 with Sc and Zr

The 7075 aluminium with 0.5 wt.% addition of combined Sc and Zr (7075ScZr) was cast in the form of rods of 30 mm diameter and 100 mm length into a copper die. The chemical composition of the alloy was determined to be: Mg- 2,83%, Cu-1,72%, Zn-5,86%, Zr, Sc – 0,5%, Al-88,73% (all in wt.%). The addition of 0.5 % of scandium and zirconium to Al-Mg-Cu-Zn alloys should cause the formation of ZrAl₃ and ScAl₃ phases. Crystallization nuclei based on Al₃ (Zr, Sc) phase play a significant role in refining of grains [15]. The curves in Figure 6 show the dependence of the amount of liquid phase as a function of temperature for the 7075ScZr calculated from the DSC measurements for heating. It shows that Al(α) melting started at 565°C for heating rate 15°C/min. The end of melting took place at 657°C for heating from the solid state. The semi-solid processing was carried out at 632°C, which corresponds to 25% of liquid phase in accordance with the DSC analysis (point A marked in Fig 6).

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5.1. Analysis of thixo-formed 7075ScZr aluminium alloy

The optical microstructure of a thixo-formed part is shown in Figure 7. The microstructure was homogeneous, and consisted of globular grains surrounded by the eutectic mixture. The average size of α (Al) was 26 μ m.



Fig 7 Microstructure of the thixo-casts made of steel 7075ScZr [15]

The precipitates of M ((Cu, Zn, Al)2Mg)) and T ((Cu, Zn, Al)49Mg32)) phases are present among the grains [10]. Amount of the eutectic mixture and precipitates of M and T phases was estimated from 28 to 32% [10].

6. Conclusions

Strain Induce and Melting Activated for steels as well as modification of molten 7075 aluminium alloy with 0.5% Sc and Zr appeared to be an effective methods of feedstock preparation for the thixoforming process. Suitable determination of solidus-liquidus range and technological parameters of process enable to obtain thixo-cast with defects.

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